With the beginning of the crime scene investigation (CSI) shows on TV, people started to show curiosity and interest in this field. West Virginia University was one of the leaders in starting a forensic chemistry major in their program. After 9/11/2001, the demand for forensic investigators was overwhelming which urged more universities to start this major in their programs. I though it would be a good idea to offer a non-science major course that fulfills the requirements for chemistry or a science course and at the same time it would be an interesting topic for a lot of non-science major students.

**Text Book:** "Investigating Chemistry, A Forensic Science Perspective", by Matthew E. Johll, 3rd Edition, Freeman

**Homework:** Sapling Learning [https://www.saplinglearning.com/ibiscms/login/](https://www.saplinglearning.com/ibiscms/login/) or http://www.macmillanlearning.com/Catalog/elearningbrowsebymediatype/SaplingLearning

**Course Description**

This is a 3 credit course which is designed for the non-science major students to stimulate their interest in the forensic chemistry and help them appreciate and understand the basic fundamental concepts of chemistry. In each chapter, chemical concepts related to a forensic topic are introduced in addition to a brief description of an analytical instrumentation or methodology used in crime investigation and a case study.

The main purpose of this course is to deliver the chemistry concepts to students without going into great details.

**Course Objectives**

By the end of the semester, it is expected that the student should have a clear idea of what forensic chemistry is all about including definition, history, sub-disciplines, evidence handling, reliable analytical methods and accurate data, critical thinking and scientific approaches in crime investigation in addition to the basic fundamental concepts of general chemistry. Student should have a general idea about the analytical instrumentation used in a forensic lab.

**Honor System**

The Georgetown University Honor Pledge:

*In the pursuit of the high ideals and rigorous standards of academic life,*
I commit myself to respect and uphold the Georgetown University Honor System:
To be honest in any academic endeavor, and
To conduct myself honorably, as a responsible member of the Georgetown community, as we live and work together.

You are responsible for familiarizing yourself with the Georgetown University Honor System. Information can be found at: http://www.georgetown.edu/undergrad/bulletin/regulations6.html

Classroom Conduct

In this course, as well as in all other courses, the academic policies and conduct of Georgetown University are applied. I am committed to maintaining a classroom environment free of harassment and discrimination. I value different backgrounds and communication styles and I ask that all of you contribute to making a high standard of classroom civility by being respectful of your peers, your instructor and the regulations outlined in this syllabus. The use of cell phones, PDA’s, laptop computers, etc. is not allowed during lectures. Eating or drinking in the classroom is prohibited by Georgetown University’s policy.

Course Structure

Attendance is mandatory and counts 10% of the final grades. Please inform me if you have an excuse. One unexcused absence will cost you 2%. Two unexcused absence will cost you 5%. Three unexcused absence will cost you 10%.

Problem Sets: There will be assignments of problem sets for each chapter which will help you tremendously in the exams. No grades on this set of problems

Exams: There will be 3 x 60 min exams on June 10, June 18, and June 26. Each exams counts 20% of the final grades

Sapling Homework: The Sapling homework counts for 20% which will replace the lowest score exam

Final Exam: The final exam is a comprehensive and cumulative exam which counts 30% of the final grades and will be given on July 3rd on the last class session.

Letter grades are determined based on your cumulative total raw score during the semester. The letter grade equivalents are as follows:

<table>
<thead>
<tr>
<th>Raw score</th>
<th>Letter grade</th>
<th>Raw score</th>
<th>Letter grade</th>
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<tbody>
<tr>
<td>92.5-100</td>
<td>A</td>
<td>72.5-77.4</td>
<td>C</td>
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<tr>
<td>89.5-92.4</td>
<td>A-</td>
<td>69.5-72.4</td>
<td>C-</td>
</tr>
<tr>
<td>87.5-89.4</td>
<td>B+</td>
<td>67.5-69.4</td>
<td>D+</td>
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<td>D</td>
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<td>B-</td>
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<tr>
<td>77.5-79.4</td>
<td>C+</td>
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</tbody>
</table>
Subjects to Be Covered

Chapter I  Introduction to Forensic Chemistry
Chapter II  Evidence, Collection and Preservation
Chapter III  Atomic Clues
Chapter IV  Chemical Evidence
Chapter V  Chemistry of Bonding: Structure and Function of Drug Molecules
Chapter VI  Properties of Solutions I: Aqueous Solutions
Chapter VII  Properties of Solutions II: Intermolecular Forces and Colligative Properties
Chapter VIII  Drug Chemistry
Chapter IX  Arson Investigation
Chapter X  Chemistry of Explosions
Chapter XI  Estimating the Time of Death
Chapter XII  The Nuclear Age: Energy, Medicine, and Terrorism
Chapter XIII  Poisons
Chapter IXV  Identification of Victims: DNA Analysis

Dates to Remember

<table>
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<tr>
<th>Event</th>
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<tbody>
<tr>
<td>Classes Begin</td>
<td>6/3/2019</td>
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<tr>
<td>Classes End</td>
<td>7/3/2019</td>
</tr>
<tr>
<td>Last Day to Add/Drop</td>
<td>6/5/2019</td>
</tr>
<tr>
<td>Last Day for Pass/Fail</td>
<td>6/5/2019</td>
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<tr>
<td>Last Day for Withdrawal</td>
<td>6/24/2019</td>
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<tr>
<td>Undergraduate Grades Due</td>
<td>7/12/2019</td>
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</table>

Looking forward to working with you and good luck!